## **Friday Worksheet**

## **Chromatography 1**

1) Two different food dye samples, W and Z were compared using thin layer chromatography as shown below.



- a) Calculate the  $R_f$  value of W and Z W = 8.0/9.1 = 0.88, Z = 2.5/9.1 = 0.28
- b) Which dye is more strongly adsorbed?
  Z
- c) Explain how dyes W and Z can be identified? Under these conditions they have a unique R<sub>f</sub> value which can be used to identify these compounds
- d) On the right is another chromatogram of a food substance containing food dye. Is dye "W" present? Explain.
   Yes

C has an  $R_f$  value of 0.88 same as W. If each chromatogram was run under identical conditions then the  $R_f$  value is identical to W.



